



A New, General and Regioselective Method for the Synthesis of 2,6-Disubstituted 4-Aminopteridines

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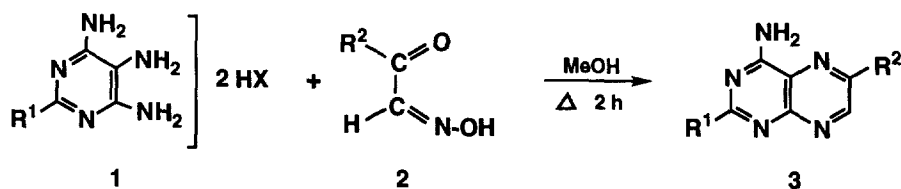
Abstract: A new synthetic method for the preparation of 2,6-disubstituted 4-aminopteridines is described. Treatment of 2-substituted 4,5,6-triaminopyrimidines with α -ketoaldoximes in MeOH affords in a regioselective one-step reaction 2,6-disubstituted 4-aminopteridines in high yield.
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The classical synthetic route to pteridines was developed by Isay and Gabriel¹ and involves condensations of 5,6-diaminopyrimidines with 1,2-dicarbonyl compounds. Syntheses using symmetrical 1,2-dicarbonyl derivatives present no regioselectivity problems, but unsymmetrical dicarbonyl components give 6- and 7-substituted product mixtures of ambiguous structures. In special cases, pH-variations effect the orientation¹ but a defined regioselective synthesis affords usually an unambiguous approach as demonstrated by the Timmis¹, the Polonovski-Boon¹ and the Taylor¹ reaction, respectively. Since the majority of naturally occurring pteridines possess a side-chain in 6-position special interest in the development of a simple synthetic approach leading to 6-substituted pteridines even in technical scale is obvious.

Our new method works especially well for the regioselective synthesis of 2,6-disubstituted 4-aminopteridines and is derived from an observation by Kang, Soyka and Pfeleiderer² using oximinoacetone for the preparation of 1,3,6-trimethylumazine in a two-step condensation reaction starting from 5,6-diamino-1,3-dimethyluracil.

The general application of the regioselective synthesis could be demonstrated by the condensation of 2-substituted 4,5,6-triaminopyrimidine dihydrohalides (**1**; X=Br, Cl) with α -ketoaldoximes **2** to form 2,6-disubstituted 4-aminopteridines (**3**) in good to excellent yields (scheme 1). We are able to prepare the 6-substituted pteridines derivatives **3a-t** in a facile one-step reaction with high purity⁶. The 6-halomethyl-pteridines **3b, c, h, l, m, q, r** are especially valuable intermediates from a synthetic point of view due to easy nucleophilic substitution reactions in the side-chain leading to folic acid analogues such as the important anti-cancer agents aminopterin and methotrexate, respectively³.

Additionally a great synthetic potential can be attributed to the 6-substituted 2,4-diaminopteridines which are easily converted into the corresponding pterins by either alkaline or acid hydrolysis of the 4-amino-group offering even more flexibility of the new approach⁵.



3	X	R ¹	R ²	Yield [%]	3	X	R ¹	R ²	Yield [%]
a	Cl	NH ₂	CH ₃	81	k	Cl	SCH ₃	CH ₃	47
b	Br	NH ₂	CH ₂ Br	88	l	Br	SCH ₃	CH ₂ Br	61
c	Cl	NH ₂	CH ₂ Cl	79	m	Cl	SCH ₃	CH ₂ Cl	74
d	Cl	NH ₂	CH ₂ CH ₃	76	n	Cl	SCH ₃	C ₆ H ₅	97
e	Cl	NH ₂	C ₆ H ₅	93	o	Cl	SCH ₃	p-CH ₃ -C ₆ H ₄	97
f	Cl	NH ₂	p-CH ₃ -C ₆ H ₄	88	p	Cl	C ₆ H ₅	CH ₃	61
g	Cl	H	CH ₃	31	q	Br	C ₆ H ₅	CH ₂ Br	73
h	Cl	H	CH ₂ Cl	66	r	Cl	C ₆ H ₅	CH ₂ Cl	87
i	Cl	H	C ₆ H ₅	88	s	Cl	C ₆ H ₅	C ₆ H ₅	88
j	Cl	H	p-CH ₃ -C ₆ H ₄	90	t	Cl	C ₆ H ₅	p-CH ₃ -C ₆ H ₄	89

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REFERENCES AND NOTES

- For general references on pteridine chemistry: Pfeleiderer, W. in "Comprehensive Heterocyclic Chemistry", Katritzky A.R ; Rees C. W., eds, Vol 3, Part 2B, Pergamon Press, Oxford, 1984, p 263-327.
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- ¹H NMR (250 MHz, ppm, DMSO-d₆), δ (s, 1H, C(7)-H): **3a**: 8.60, **3b**: 8.84, **3c**: 8.84, **3d**: 8.63, **3e**: 9.33, **3f**: 9.31, **3g**: 8.97, **3h**: 9.21, **3i**: 9.69, **3j**: 9.68, **3k**: 8.85, **3l**: 9.09, **3m**: 9.09, **3n**: 9.61, **3o**: 9.57, **3p**: 8.97, **3q**: 9.21, **3r**: 9.21, **3s**: 9.74, **3t**: 9.70.
- The results of these studies have been applied for a patent by LONZA AG, Visp-Basel / Switzerland Pfeleiderer, W.; Taghavi-Moghadam, Schweiz. Patentgesuch-No. 03503/95.
- Typical procedure** : A suspension of 5 mmol 2-substituted 4,5,6-triaminopyrimidine dihydrohalide (**1**; X=Br, Cl) in 50 ml methanol was treated with a solution of 7.5 mmol of an α-ketoaldoxime (**2**) in 10 ml of MeOH at reflux temperature for 2h. The corresponding reaction products (**3**) were collected after neutralisation with conc. NH₃ at room temperature, washed with methanol, ether and dried at 100 °C in an oven. The isomer purity of the isolated products was proven by ¹H NMR spectroscopy since the chemical shift of the H-C(7) in the pteridine system is decisive for the structural assignments of the isomers⁴.

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